

(19)



Europäisches Patentamt  
European Patent Office  
Office européen des brevets



(11)

**EP 1 211 300 A2**

(12)

**EUROPEAN PATENT APPLICATION**

(43) Date of publication:

**05.06.2002 Bulletin 2002/23**

(51) Int Cl.7: **C09K 9/00**

(21) Application number: **01128409.8**

(22) Date of filing: **04.12.2001**

(84) Designated Contracting States:

**AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU  
MC NL PT SE TR**

Designated Extension States:

**AL LT LV MK RO SI**

(30) Priority: **04.12.2000 JP 2000368015**

**15.10.2001 JP 2001316342**

(71) Applicant: **THE PILOT INK CO., LTD.**

**Nagoya-shi, Aichi-ken (JP)**

(72) Inventors:

- **Senga, Kuniyuki**  
**Showa-ku, Nagoya-shi, Aichi (JP)**
- **Sugai, Jun**  
**Showa-ku, Nagoya-shi, Aichi (JP)**
- **Koide, Shigehiro**  
**Showa-ku, Nagoya-shi, Aichi (JP)**

(74) Representative: **HOFFMANN - EITLE**

**Patent- und Rechtsanwälte**  
**Arabellastrasse 4**  
**81925 München (DE)**

(54) **Reversibly thermochromic light-transmitting laminate member**

(57) A reversibly thermochromic light-transmitting laminate member which has a satisfactory thermochromic function and undergoes alternate color changes between a colorless transparent state and a colored transparent state.

**EP 1 211 300 A2**

## Description

### FIELD OF THE INVENTION

[0001] The present invention relates to a reversibly thermochromic light-transmitting laminate member. More particularly, the invention relates to a reversibly thermochromic light-transmitting laminate member comprising a substrate and a reversibly thermochromic layer formed thereon, which undergoes alternate color changes between a colorless transparent state and a colored transparent state with temperature.

### BACKGROUND OF THE INVENTION

[0002] A laminate member which undergoes alternate color changes between a colorless transparent state and a colored transparent state with temperature is disclosed in U.S. Patent No. 5,721,059. This laminate member comprises a substrate and formed thereon a reversibly thermochromic layer comprising a vinyl chloride/vinyl acetate copolymer resin matrix and dispersed therein a reversibly thermochromic composition comprising (a) an electron-donating color-forming organic compound, (b) an electron-accepting compound, and (c) a reaction medium determining the temperatures at which the color reactions of component (a) with component (b) take place.

[0003] The thermochromic composition contained in the reversibly thermochromic layer is not encapsulated, i.e., of the so-called non-capsule type. Because of this, not only the thermochromic layer has a factor contributing to chemical instability, but also the substrate is limited in material. With respect to thermochromic properties, the laminate member has such color-memorizing properties that it is wide hysteric and memorizes and retains a color-changed state resulting from a temperature change.

[0004] Furthermore, a composition which contains a microencapsulated pigment containing a reversibly thermochromic composition and has excellent transparency in a decolored state is disclosed in JP-A-3-290467 (the term "JP-A" as used herein means an unexamined published Japanese patent application). This composition is a thermochromic resin composition containing a microencapsulated pigment which has a particle diameter of 1  $\mu\text{m}$  or smaller and which itself has light transmission.

[0005] This resin composition is intended to have transparency imparted thereto by the light transmission of the pigment itself. However, the small capsule particle diameter may cause troubles concerning thermochromic function, e.g., difficulties in determining a color change temperature.

### SUMMARY OF THE INVENTION

[0006] An aim of the invention is to eliminate the dis-

advantages of the reversibly thermochromic laminate member which undergoes alternate color changes between a colorless transparent state and a colored transparent state. Namely, the aim is to provide a reversibly thermochromic light-transmitting laminate member which has a satisfactory thermochromic function and is free from the factor contributing to chemical instability inherent in the system employing a non-capsule type reversibly thermochromic composition.

[0007] The invention provides a reversibly thermochromic light-transmitting laminate member which comprises a substrate and formed thereon a reversibly thermochromic layer comprising a transparent resin and fixedly dispersed therein a reversibly thermochromic microencapsulated pigment containing a reversibly thermochromic composition comprising (a) an electron-donating color-forming organic compound, (b) an electron-accepting compound, and (c) a reaction medium determining the temperatures at which the color reactions of component (a) with component (b) take place, the reversibly thermochromic microencapsulated pigment being a deformable pigment which has an average particle diameter of from 1.0 to 5.0  $\mu\text{m}$ , the transparent resin having a tensile elongation at break of 400% or higher. In the laminate member of the invention, the proportion of the reversibly thermochromic microencapsulated pigment to the transparent resin is preferably from 1/0.5 to 1/20 (in terms of weight ratio on a solid basis) and the reversibly thermochromic layer preferably has a thickness of from 1 to 100  $\mu\text{m}$ . Furthermore, the transparent resin is preferably a resin selected from the group consisting of ethylene/vinyl acetate copolymer resins, polyurethane resins, and polyamide resins. In one embodiment of the invention, the laminate member has, formed over the reversibly thermochromic layer, a metallic glossy layer containing a metallic luster pigment having transparency. This metallic glossy layer preferably has a light transmittance of 5% or higher.

### DETAILED DESCRIPTION OF THE INVENTION

[0008] The laminate member of the invention, which undergoes alternate color changes between a colorless transparent state and a colored transparent state with temperature, is obtained based on a combination of a reversibly thermochromic microencapsulated pigment having an average particle diameter of from 1.0 to 5.0  $\mu\text{m}$  with a transparent resin having a tensile elongation at break of 400% or higher, as stated above.

[0009] The average particle diameter of the microcapsule pigments can be measured as follows. 1 part of the microcapsule pigment and 10 parts of water are put into a 20 ml-capacity glass vessel, the mixture was subjected to the dispersing treatment for 15 minutes using an ultrasonic dispersing machine, and the average particle diameter of the sample was measured in a cell of a laser diffraction and scattering particle size distribution analyser (LA-300, manufactured by Horiba Ltd.).

[0010] The tensile elongation at break can be measured in accordance with JIS K6732. A test piece which has a suitable shape for the tensile testing is prepared from the transparent resin and the tensile elongation at break is measured using a tensile tester having a constant rate of the crosshead travel (Autograph AGS, Shimadzu Corporation) at  $23^{\circ}\text{C} \pm 2^{\circ}\text{C}$  with the rate of pulling of  $200 \pm \text{mm/min}$ .

[0011] More particularly, the reversibly thermochromic layer comprising a combination of a microencapsulated pigment having an average particle diameter of from  $1.0$  to  $5.0 \mu\text{m}$  with a transparent resin having a tensile elongation at break of  $400\%$  or higher is presumed to undergo the following phenomena.

[0012] The microencapsulated pigment has different capsule shapes depending on whether the reversibly thermochromic composition contained therein is in a solid state, i.e., a colored state, or is in a dissolved state, i.e., a decolored state. Specifically, when the encapsulated composition is in a dissolved state (decolored state), then the capsules have a degree of freedom and are deformable because the reversibly thermochromic composition is liquid. However, when the encapsulated composition is in a solid state (colored state), the capsules have recesses in the surface thereof, which were formed upon solidification of the reversibly thermochromic composition, and are hard and less apt to be deformed.

[0013] On the other hand, the transparent resin used in the invention has an elongation as high as  $400\%$  or above and is capable of conforming to capsule shape changes with temperature. Namely, at any time when the capsules change in shape with temperature, the resin surrounding the capsules is capable of deforming so as to conform to the capsule shapes. As a result, formation of spaces between the capsule surface and the resin is inhibited and a bleaching phenomenon can be prevented. Consequently, the reversibly thermochromic layer can undergo alternate color changes between a colorless transparent state and a colored transparent state with temperature over long without impairing the light transmission of the microencapsulated pigment.

[0014] Requirements for enabling the color changes to occur include the particle diameter of the capsules which is regulated to  $1.0$  to  $5.0 \mu\text{m}$ . The reasons for this are as follows. In case where the particle diameter of the capsules is smaller than  $1.0 \mu\text{m}$ , not only satisfactory thermochromic properties cannot be obtained, but also such capsules should have a reduced wall thickness and hence have poor durability in secondary processing, making it difficult to satisfy suitability for practical use. In case where the particle diameter thereof exceeds  $5.0 \mu\text{m}$ , the microencapsulated pigment itself has reduced light transmission and, hence, the reversibly thermochromic layer in a colored state has reduced light transmission. The particle diameter of the capsules is preferably from  $1.0$  to  $3.0 \mu\text{m}$ , more preferably from  $1.0$  to  $2.0 \mu\text{m}$ . Use of such capsules enables the reversibly

thermochromic layer to show transparency and a sufficient color density in a colored state, be reduced in a residual color in a decolored state, and have excellent sharpness in color change. Namely, the reversibly thermochromic layer can have an excellent balance among thermochromic functions.

[0015] The proportion of the microencapsulated pigment to the transparent resin in terms of weight ratio on a solid basis is preferably from  $1/0.5$  to  $1/20$ , more preferably from  $1/1$  to  $1/10$ . The thickness of the reversibly thermochromic layer is preferably from  $1$  to  $100 \mu\text{m}$ , more preferably from  $10$  to  $80 \mu\text{m}$ . In case where the proportion of the microencapsulated pigment to the transparent resin in terms of weight ratio on a solid basis exceeds  $1/0.5$ , transparency is reduced and the reversibly thermochromic coating layer has a reduced strength. In case where the proportion thereof is lower than  $1/20$ , i.e., the proportion of the transparent resin exceeds the upper limit, then an excessively large film thickness is necessary for obtaining a satisfactory color density and this leads to insufficient drying and to problems concerning coating period, etc. Thus, such too high proportions of the transparent resin are unsuitable for practical use.

[0016] Thicknesses of the reversibly thermochromic layer smaller than  $1 \mu\text{m}$  result in difficulties in obtaining a satisfactory color tone in a colored state. On the other hand, thicknesses thereof exceeding  $100 \mu\text{m}$  may impair the transparency of the layer.

[0017] The reversibly thermochromic composition comprises three ingredients, i.e., (a) an electron-donating compound, (b) an electron-accepting compound, and (c) a reaction medium determining the temperatures at which the color reactions between components (a) and (b) take place. As this composition can be used, for example, the compositions proposed by the applicant which are disclosed in U.S. Patents No. 4,028,118 and No. 4,732,810.

[0018] This composition changes in color at a given temperature (color change point) in such a manner that the composition is in a decolored state at temperatures not lower than the color change point and is in a colored state at temperatures lower than the color change point. At ordinary temperature, the composition exists only in specific one of the two states. Namely, the other state is maintained only during the period in which heat or cold required for the development of this state is kept being applied, and the composition returns to the original ordinary-temperature state upon removal of the heat or cold. This composition is a reversibly thermochromic composition of the heating decoloration type which shows such a thermochromic behavior in which the hysteresis range is relatively narrow.

[0019] Also usable is the thermochromic composition which is disclosed in U.S. Patent No. 4,720,301. This composition shows enhanced hysteresis. Namely, plotting the color density against temperature gives a curve showing that the composition changes in color along

routes which differ considerably between the case where the composition is heated from a temperature on the lower-temperature side of the color change temperature range and in the reverse case where the composition is cooled from a temperature on the higher-temperature side of the color change temperature range. This composition is a thermochromic color-memorizing composition of the heating decoloration type in which the colored state in the range of temperatures not higher than the lower color change point and the decolored state in the range of temperatures not lower than the higher color change point can be alternately memorized and retained even after the heat or cold required for the color change has been removed.

**[0020]** Furthermore, a reversibly thermochromic composition of the heating coloration type can also be used which contains as an electron-accepting compound (b) a specific alkoxyphenol compound having a linear or branched alkyl group having 3 to 18 carbon atoms.

**[0021]** The reversibly thermochromic composition is microencapsulated and used as a microencapsulated pigment. This microencapsulation enables the composition to develop a bright color having a high density and satisfy practical properties such as homogeneity, dispersion stability, chemical resistance, and heat resistance.

**[0022]** For the microencapsulation, known techniques can be used. Examples thereof include interfacial polymerization, in-situ polymerization, in-liquid curing coating, phase separation from an aqueous solution, phase separation from an organic solvent, melt dispersion cooling, air-suspension coating, and spray drying.

**[0023]** Before the microencapsulated pigment is subjected to practical use, the surface properties of the microcapsules may be modified according to purposes.

**[0024]** The microencapsulated pigment is dispersed into a medium containing the transparent resin to prepare a coloring material such as an ink or coating composition. This coloring material is applied to a substrate to thereby form a reversibly thermochromic layer.

**[0025]** Adding a non-thermochromic dye or pigment enables a color change from a colored transparent state (1) to another colored transparent state (2).

**[0026]** For forming the thermochromic layer, known techniques can be used. Examples thereof include printing techniques such as screen printing, offset printing, gravure printing, coater printing, tampon printing, and transfer printing and coating techniques such as brush coating, spray coating, electrostatic coating, electrodeposition, curtain coating, roller coating, and dip coating.

**[0027]** Examples of the material of the substrate include papers, synthetic papers, artificial leathers, natural leathers, plastics, glasses, porcelains and pottery, woods, stones, and metals.

**[0028]** The substrate may have a flat shape or may have recesses and protrusions.

**[0029]** Since the reversibly thermochromic layer of

the laminate member of the invention undergoes alternate color changes between a colorless transparent state and a colored transparent state with temperature, it is preferred that the substrate be made of a transparent material. This is because in the laminate member employing a transparent substrate, the condition on the opposite side of the substrate can be perceived regardless of whether the thermochromic layer is in a colorless transparent state or a colored transparent state. This laminate member can be applied, for example, to a reversibly thermochromic three-dimensional object comprising a three-dimensional substrate made of a transparent resin, the reversibly thermochromic layer formed on the substrate, and an appropriate shaped article enclosed in the three-dimensional substrate. In this reversibly thermochromic object, not only the shaped article placed inside can be always perceived but also the thermochromic object can have such unexpected color-changing properties that the reversibly thermochromic layer undergoes color changes with temperature.

**[0030]** Examples of the shaped article include dolls and toys modeled on, e.g., animals, vehicles, buildings, plants, foods, or stones.

**[0031]** The shaped article enclosed in the three-dimensional object is preferably in close contact with the three-dimensional object. Namely, the reversibly thermochromic three-dimensional object preferably has such a constitution that it has no space at the interface between the outer surface of the shaped article and the three-dimensional object. This is because the absence of spaces enables the shaped article to be clearly perceived.

**[0032]** The substrate may be one having an optical property selected from iridescence, suitability for use as a hologram-recording material, metallic luster, pearlescence, and fluorescence. In this constitution, not only the optical property can be always perceived but also the layered object can have such unexpected color-changing properties that the reversibly thermochromic layer undergoes color changes with temperature.

**[0033]** Examples of materials having those optical properties include: a film bearing an optical interference pattern and comprising one or more of oxides, sulfides, and fluorides; an iridescent, transparent, multilayer film causing light interference which has a hundred or more interlayers made of two or more polymers differing in refractive index; a hologram film formed by pressing a hologram master bearing a hologram against any of various resins to obtain an embossed film and then vapor-depositing a metal such as Al, Cr, Ni, Sn, Fe, Co, Cu, Pb, Sb, Mg, Cd, or Bi on the embossed resin layer in a thickness of from 200 to 2,000 Å; metal pieces; and materials obtained by suitably adhering a metal foil or a pearlescent pigment, e.g., a mica coated with titanium dioxide, mica coated with iron oxide and titanium dioxide, mica coated with iron oxide, guanine, sericite, basic lead carbonate, acid lead arsenate, or bithmuth oxychloride, to a surface of a support.

[0034] The metallic glossy layer which may be formed on the reversibly thermochromic layer will be explained next.

[0035] The metallic glossy layer is a layer comprising a binder resin and, fixedly dispersed therein, a metallic luster pigment based on a transparent material, as a core material, selected from natural micas, synthetic micas, flat pieces of glasses, and flaky aluminum oxide.

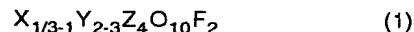
[0036] The metallic glossy layer not only imparts any of various colors with a metallic luster to the reversibly thermochromic layer, but also functions as a light-shielding layer. Namely, due to its light-absorbing (or light-reflecting) function and light-shielding function, the metallic glossy layer absorbs or reflects at least part of ultra-violet and visible rays, which may adversely influence the functions of the reversibly thermochromic layer. Thus, the metallic glossy layer further has the effect of improving the light resistance of the reversibly thermochromic layer.

[0037] Effective metallic luster pigments based on a natural mica as a core material include one obtained by coating the surface of natural-mica particles with titanium oxide, one obtained by coating the surface of natural-mica particles with titanium oxide and then further coating the coated particles with iron oxide, and one obtained by coating the surface of natural-mica particles with titanium oxide and then further coating the coated particles with a non-thermochromic dye or pigment (preferably in an amount of from 0.5 to 10% by weight). Specific examples thereof include: a gold metallic luster pigment having a particle size of from 5 to 50  $\mu\text{m}$  obtained by coating the surface of a natural mica with 41 to 44% by weight titanium oxide; a gold metallic luster pigment having a particle size of from 5 to 60  $\mu\text{m}$  obtained by coating the surface of a natural mica with 30 to 38% by weight titanium oxide and further coating the coated particles with 0.5 to 10% by weight non-thermochromic color pigment; a silver metallic luster pigment having a particle size of from 5 to 100  $\mu\text{m}$  obtained by coating the surface of a natural mica with 16 to 39% by weight titanium oxide; a metallic luster pigment obtained by coating the surface of a natural mica with 45 to 58% by weight titanium oxide; and a metallic luster pigment obtained by coating the surface of a natural mica with 45 to 58% by weight titanium oxide and then further coating the coated particles with 0.5 to 10% by weight non-thermochromic color dye or pigment. The "% by weight" used above is based on the total weight of the metallic luster pigment.

[0038] Metallic luster pigments based on a synthetic mica as a core material have a lower impurity content and a lower content of iron and other metal ions, which are a coloring factor, as compared with the metallic luster pigments based on a natural mica. Because of this, the pigments based on a synthetic mica have excellent transparency and is highly lustrous and glittering. Effective examples thereof include a pigment which is obtained by coating the surface of a synthetic mica with

one or more metal oxides comprising titanium oxide as the main component and has an average particle thickness of from 0.1 to 5  $\mu\text{m}$  and an average particle diameter of from 2 to 1,000  $\mu\text{m}$ . This pigment has a metallic luster of a gold, silver, or another color depending on the degree of coating with the metal oxides. It is so transparent that the color tone of the underlying reversibly thermochromic layer can be clearly perceived.

[0039] Examples of the synthetic mica include ones represented by the following formula. One example is  $\text{KMg}_3(\text{AlSi}_3\text{O}_{10})\text{F}_2$ .



[0040] (In formula (1), X represents  $\text{Na}^+$ ,  $\text{Li}^+$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Sr}^{2+}$ , or  $\text{Ba}^{2+}$ ; Y represents one or more members selected from  $\text{Mg}^{2+}$ ,  $\text{Li}^+$ ,  $\text{Ni}^{2+}$ ,  $\text{Co}^{2+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Mn}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Cr}^{3+}$ , and  $\text{Ti}^{3+}$ ; and Z represents one or more members selected from  $\text{Al}^{3+}$ ,  $\text{Si}^{4+}$ ,  $\text{Ge}^{4+}$ , and  $\text{B}^{3+}$ .)

[0041] The shape of the mica is not particularly limited, and examples thereof include flat shapes and flaky shapes.

[0042] Examples of the metal oxides with which the surface of the synthetic mica is coated include oxides of metals such as titanium, zirconium, chromium, vanadium, and iron. However, it is preferred to use one or more metal oxides comprising titanium oxide as the main component.

[0043] The metallic luster pigments based on a synthetic mica as a core material are effective when the average particle thickness thereof is generally from 0.1 to 5  $\mu\text{m}$  and the average particle diameter thereof is generally from 2 to 1,000  $\mu\text{m}$ , preferably from 2 to 500  $\mu\text{m}$ , more preferably from 2 to 200  $\mu\text{m}$ . This average particle diameter means the average particle diameter which is determined by laser diffractometry and is the particle diameter corresponding to the 50% point in a cumulative median diameter distribution on a volume basis.

[0044] Specific examples of the metallic luster pigments comprising a synthetic mica whose surface is coated with a metal oxide include "Ultimica" (trade name) SB-100 (5-30  $\mu\text{m}$ ; silver), SD-100 (10-60  $\mu\text{m}$ ; silver), SE-100 (15-100  $\mu\text{m}$ ; silver), SF-100 (44-150  $\mu\text{m}$ ; silver), SH-100 (150-600  $\mu\text{m}$ ; silver), YB-100 (5-30  $\mu\text{m}$ ; gold), YD-100 (10-60  $\mu\text{m}$ ; gold), YE-100 (15-100  $\mu\text{m}$ ; gold), YF-100 (44-150  $\mu\text{m}$ ; gold), RB-100 (5-300  $\mu\text{m}$ ; metallic red), RD-100 (10-60  $\mu\text{m}$ ; metallic red), RE-100 (15-100  $\mu\text{m}$ ; metallic red), RF-100 (44-150  $\mu\text{m}$ ; metallic red), RBB-100 (5-30  $\mu\text{m}$ ; metallic purple), RBD-100 (10-60  $\mu\text{m}$ ; metallic purple), RBE-100 (15-100  $\mu\text{m}$ ; metallic purple), RBF-100 (44-150  $\mu\text{m}$ ; metallic purple), VB-100 (5-30  $\mu\text{m}$ ; metallic violet), VD-100 (10-60  $\mu\text{m}$ ; metallic violet), VE-100 (15-100  $\mu\text{m}$ ; metallic violet), VF-100 (44-150  $\mu\text{m}$ ; metallic violet), BB-100 (5-30  $\mu\text{m}$ ; metallic blue), BD-100 (10-60  $\mu\text{m}$ ; metallic blue), BE-100 (15-100  $\mu\text{m}$ ; metallic blue), BF-100 (44-150  $\mu\text{m}$ ; metallic

blue), GB-100 (5-30  $\mu\text{m}$ ; metallic green), GD-100 (10-60  $\mu\text{m}$ ; metallic green), GE-100 (15-100  $\mu\text{m}$ ; metallic green), and GF-100 (44-150  $\mu\text{m}$ ; metallic green), all manufactured by Nihon Kokenkogyo K.K.

**[0045]** Examples of the metallic luster pigments obtained by coating the surface of flat glass pieces as a core material with titanium oxide include such metallic luster pigments which have a gold, silver, or another metallic color depending on the degree of coating with titanium oxide. Specific examples thereof include the pigments obtained by coating glass flakes with titanium oxide which are manufactured by Nippon Sheet Glass Co., Ltd. and sold under the trade name of "Metashine" RCF-SX-5450TS(6041) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $450\pm 145$   $\mu\text{m}$ ; gold), RCFSX-5200TS (6042) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $200\pm 70$   $\mu\text{m}$ ; silver), RCFSX-5140TS(6043) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $140\pm 45$   $\mu\text{m}$ ; silver), RCFSX-5080TS(6044) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $80\pm 30$   $\mu\text{m}$ ; silver), RCF-SX-2080TS(6046) (average thickness,  $2\pm 1$   $\mu\text{m}$ ; average particle size,  $80\pm 30$   $\mu\text{m}$ ; silver), RCFSX-K120TS (6043) (average thickness,  $20\pm 5$   $\mu\text{m}$ ; average particle size,  $120\pm 20$   $\mu\text{m}$ ; silver), RCFSX-5090RC(8052) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $90\pm 30$   $\mu\text{m}$ ; gold), RCFSX-5090RC(8053) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $90\pm 30$   $\mu\text{m}$ ; metallic green), RCFSX-5090RC(8069) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $90\pm 30$   $\mu\text{m}$ ; metallic blue), RCFSX-5090RC(8070) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $90\pm 30$   $\mu\text{m}$ ; metallic purple), RCF-SX-5090RC(8071) (average thickness,  $5\pm 2$   $\mu\text{m}$ ; average particle size,  $90\pm 30$   $\mu\text{m}$ ; metallic red), RCFSX-1040RC(9543) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $40$   $\mu\text{m}$ ; silver), RCFSX-1040RC(9544) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $40$   $\mu\text{m}$ ; gold), RCFSX-1040RC(9546) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $40$   $\mu\text{m}$ ; metallic red), RCFSX-1040RC (9548) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $40$   $\mu\text{m}$ ; metallic blue), RCFSX-1040RC(9549) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $40$   $\mu\text{m}$ ; metallic green), RCFSX-1020RC(9550) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $20$   $\mu\text{m}$ ; silver), RCFSX-1020RC(9551) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $20$   $\mu\text{m}$ ; gold), RCFSX-1020RC(9553) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $20$   $\mu\text{m}$ ; metallic red), and RCFSX-1020RC(9555) (average thickness,  $1$   $\mu\text{m}$ ; average particle size,  $20$   $\mu\text{m}$ ; metallic blue).

**[0046]** The metallic luster pigments obtained by coating the surface of flaky aluminum oxide as a core material with a metal oxide will be explained below.

**[0047]** Because the core material is flaky aluminum oxide, this kind of metallic luster pigment has a lower impurity content and is more lustrous than the pigments based on a natural mica.

**[0048]** Examples of the metal oxide with which the surface of the aluminum oxide is coated include oxides of metals such as titanium, zirconium, chromium, vana-

dium, and iron. However, it is preferred to use one or more metal oxides comprising titanium oxide as the main component. Examples of the metallic luster pigments include ones which have a gold, silver, or another metallic color depending on the degree of coating with the metal oxide.

**[0049]** These metallic luster pigments preferably have an average thickness of from 0.1 to 5  $\mu\text{m}$  and an average particle diameter of from 2 to 200  $\mu\text{m}$ .

**[0050]** By using a metallic luster pigment having a thickness and average particle diameter within the respective ranges shown above, it is possible to obtain a metallic glossy image which is evenly lustrous and a thermochromic image undergoing distinct color changes. This average particle diameter means the average particle diameter which is determined by laser diffractometry and is the particle diameter corresponding to the 50% point in a cumulative median diameter distribution on a volume basis.

**[0051]** Specific examples of the metallic luster pigments obtained by coating the surface of flaky aluminum oxide with a metal oxide include the pigment manufactured by Merck & Co., Inc. and sold under the trade name of "Xirallic" T50-10 (10-30  $\mu\text{m}$ ; silver).

**[0052]** The metallic luster pigments obtained by coating a core material selected from natural micas, synthetic micas, flat pieces of glasses, and flaky aluminum oxide with a metal oxide have transparency as well as a metallic luster. Of these metallic luster pigments, those obtained by coating a core material selected from synthetic micas, flat pieces of glasses, and flaky aluminum oxide with a metal oxide are suitable for use in the invention because they are superior in transparency.

**[0053]** The metallic glossy layer in the invention has a light transmittance of preferably 5% or higher, more preferably 10% or higher.

**[0054]** The light transmittance of the metallic luster layer can be measured as follows. A transparent support having a metallic luster layer thereon and a transparent support having no metallic luster layer as a control are set on a spectrophotometer (Autospectrophotometer U-3210, manufactured by Hitachi Ltd.) and the light transmittance at the wavelength from 400 nm to 700 nm was measured at room temperature (20°C). The average value of the maximum value and the minimum value is calculated as the light transmittance.

**[0055]** Such a light transmittance produces the following effect. When the reversibly thermochromic layer is in a colored state, the metallic glossy layer is transparent and a metallic lustrous color is perceived. When the reversibly thermochromic layer is in a decolored state, no metallic lustrous color is perceived and the metallic glossy layer shows excellent transparency.

**[0056]** In the case where the substrate has transparency, an article placed on the back side of the laminate member can be clearly perceived through the laminate member when the light transmittance of the laminate member itself is 3% or higher or 5% or higher at the time

when the reversibly thermochromic layer is in a colored state or in a decolored state, respectively.

**[0057]** In producing the reversibly thermochromic light-transmitting laminate member of the invention, a reversibly thermochromic ink is suitably applied to a substrate. The ink comprises a reversibly thermochromic microencapsulated pigment containing a reversibly thermochromic composition comprising (a) an electron-donating color-forming organic compound, (b) an electron-accepting compound, and (c) a reaction medium determining the temperatures at which the color reactions of component (a) with component (b) take place and a vehicle containing a transparent resin having a tensile elongation at break of 400% or higher. The resultant coating is dried to form a reversibly thermochromic layer, whereby the laminate member of the invention is obtained.

**[0058]** In the case where a metallic glossy layer is to be formed on the reversibly thermochromic layer, a metallic luster ink comprising a vehicle containing a metallic luster pigment is applied to the thermochromic layer to form the metallic glossy layer.

**[0059]** The invention will be explained below in more detail by reference to Examples, but the invention should not be construed as being limited to these Examples. In the following Examples and Comparative Examples, all parts are by weight.

#### EXAMPLE 1

##### Production of Reversibly Thermochromic Microencapsulated Pigment:

**[0060]** A reversibly thermochromic composition consisting of 3 parts of 1,2-benz-6-(N-ethyl-N-isobutylamino)fluoran as an electron-donating color-forming organic compound (a), 10 parts of 1,1-bis(4-hydroxyphenyl)-n-decane and 2 parts of 2,2-bis(4-hydroxyphenyl)hexafluoropropane as electron-accepting compounds (b), 50 parts of neopentyl stearate as a reaction medium (c), and 3 parts of diheptadecyl ketone as a color change temperature regulator was heated to dissolve the ingredients. Thereto were added 30 parts of an aromatic polyisocyanate prepolymer as a wall material and 40 parts of a co-solvent. The resultant mixture was homogenized to obtain a solution. This solution was poured into an aqueous poly(vinyl alcohol) solution with agitation to emulsify the resultant mixture so that the organic phase formed minute droplets having an average particle diameter of 1.5  $\mu\text{m}$ . An aqueous solution containing 5 parts of diethylenetriamine was added dropwise to the emulsion to encapsulate the reversibly thermochromic composition with a wall made of a polyurea resin formed from the polyisocyanate and the amine. The resultant dispersion was further heated with stirring to obtain a dispersion of a microencapsulated pigment.

**[0061]** The microencapsulated pigment was a thermochromic microencapsulated pigment which became

colorless upon heating to 36°C or higher and turned pink upon cooling to 12°C or lower. This pigment was examined for particle diameter with a laser type particle size distribution analyzer (LA-300, manufactured by Horiba Ltd.). As a result, the average particle diameter thereof was found to be 1.47  $\mu\text{m}$ .

##### Preparation of Reversibly Thermochromic Light-Transmitting Material:

**[0062]** Seven parts of the microencapsulated pigment was mixed with 75 parts of a 40% solution of a polyurethane resin (elongation at break, 500%) in toluene and MIBK, 40 parts of xylene, and 30 parts of cyclohexanone. This mixture was homogenized by stirring to obtain a reversibly thermochromic light-transmitting material (microencapsulated pigment/polyurethane resin = 1/4.3).

##### Production of Reversibly Thermochromic Light-Transmitting Laminate member:

**[0063]** A surface of a toy jewel molded from a transparent methacrylic resin and laminated on the back side with a paper coated with vapor-deposited aluminum was spray-coated as a substrate with the reversibly thermochromic light-transmitting material. The resultant coating was dried at 70°C for 30 minutes to form a reversibly thermochromic layer.

**[0064]** In the reversibly thermochromic light-transmitting laminate member thus obtained, the reversibly thermochromic layer had a thickness of about 40  $\mu\text{m}$ . When the laminate member was cooled to 12°C or lower, the reversibly thermochromic composition assumed a pink color and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the laminate member was perceived as a toy jewel having a metallic pink color attributable to a combination of the pink color of the composition and the metallic luster of the paper coated with vapor-deposited aluminum. When the laminate member was heated to 36°C or higher, the reversibly thermochromic composition was decolored and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the laminate member was perceived as a toy jewel giving a feeling of high transparency and having a silver color attributable to the paper coated with vapor-deposited aluminum.

#### EXAMPLE 2

##### Production of Reversibly Thermochromic Microencapsulated Pigment:

**[0065]** A reversibly thermochromic composition consisting of 3 parts of 3-{2-ethoxy-4-(N-ethylanilino)phe-

nyl)-3-(1-ethyl-2-methylindol-3-yl)-4-azaphthalide as an electron-donating color-forming organic compound (a), 10 parts of 1,1-bis(4-hydroxyphenyl)-n-decane and 2 parts of 2,2-bis(4-hydroxyphenyl)hexafluoropropane as electron-accepting compounds (b), 40 parts of stearyl caprate and 10 parts of cetyl caprate as reaction media (c), and 3 parts of stearamide as a color change temperature regulator was heated to dissolve the ingredients. Thereto were added 30 parts of an aromatic polyisocyanate prepolymer as a wall material and 40 parts of a co-solvent. The resultant mixture was homogenized to obtain a solution. This solution was poured into an aqueous poly(vinyl alcohol) solution with agitation to emulsify the resultant mixture so that the organic phase formed minute droplets having an average particle diameter of 2.0  $\mu\text{m}$ . An aqueous solution containing 5 parts of diethylenetriamine was added dropwise to the emulsion to encapsulate the reversibly thermochromic composition with a wall made of a polyurea resin formed from the polyisocyanate and the amine. The resultant dispersion was further heated with stirring to obtain a dispersion of a microencapsulated pigment.

**[0066]** The microencapsulated pigment was a thermochromic microencapsulated pigment which became colorless upon heating to 35°C or higher and turned blue upon cooling to 25°C or lower. This pigment was examined for particle diameter with a laser type particle size distribution analyzer (LA-300, manufactured by Horiba Ltd.). As a result, the average particle diameter thereof was found to be 1.86  $\mu\text{m}$ .

#### Preparation of Reversibly Thermochromic Light-Transmitting Material:

**[0067]** Fifteen parts of the microencapsulated pigment (in the form of aqueous dispersion having a solid content of 55%) was mixed with 100 parts of an emulsion (solid content, 40%) of a polyurethane resin (elongation at break, 800%). This mixture was homogenized by stirring to obtain a reversibly thermochromic light-transmitting material (microencapsulated pigment/polyurethane resin = 1/4.8).

#### Production of Reversibly Thermochromic Light-Transmitting Laminate member:

##### **[0068]**

A minicar body as a substrate was molded from a methacrylic resin and then wholly coated with the reversibly thermochromic light-transmitting material by spraying. The resultant coating was dried at 70°C for 30 minutes to form a reversibly thermochromic layer.

In the reversibly thermochromic light-transmitting laminate member thus obtained, the reversibly thermochromic layer had a thickness of about 20  $\mu\text{m}$ . When the laminate member was cooled to 25°C or

lower, the reversibly thermochromic composition assumed a blue color and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the interior of the minicar could be perceived through the blue body. When the laminate member was heated to 35°C or higher, the reversibly thermochromic composition was decolored and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the interior of the minicar could be perceived through the transparent body.

#### EXAMPLE 3

#### Production of Reversibly Thermochromic Light-Transmitting Laminate member with Metallic Luster:

**[0069]** A metallic luster ink obtained by evenly mixing 1 part of a metallic luster pigment (trade name, Metashine RCFSX-5090(8069); manufactured by Nippon Sheet Glass Co., Ltd.), 5 parts of a xylene solution (solid content, 50%) of an acrylic/polyol resin, 5 parts of xylene, and 1 part of an isocyanate hardener was applied by spraying to the reversibly thermochromic layer of the reversibly thermochromic light-transmitting laminate member produced in Example 2. Thus, a metallic glossy layer was formed. Thereafter, a topcoat ink obtained by evenly mixing 5 parts of a xylene solution (solid content, 50%) of an acrylic/polyol resin, 5 parts of xylene, 1 part of an isocyanate hardener, and 0.1 part of an ultraviolet absorber was applied to the metallic glossy layer by spraying. The resultant coating was dried at 70°C for 30 minutes to form a transparent topcoat layer.

**[0070]** When the laminate member thus obtained was cooled to 25°C or lower, the reversibly thermochromic composition assumed a blue color and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the interior of the minicar could be perceived through the metallic-blue body.

**[0071]** When the laminate member was heated to 35°C or higher, the reversibly thermochromic composition was decolored and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the interior of the minicar could be perceived through the transparent body.

**[0072]** The light transmittance of the metallic glossy layer itself was measured with a spectrophotometer (U-3210, manufactured by Hitachi Ltd.) and was found to be 50%. The light transmittances of the laminate member were 25% and 35% when the reversibly thermochromic composition was in a colored state and in a decolored state, respectively.



**EXAMPLE 4**

Production of Reversibly Thermochromic Light-Transmitting Laminate member with Metallic Luster:

[0073] A metallic luster ink obtained by evenly mixing 1 part of a metallic luster pigment (trade name, Ultimica BD-100; manufactured by Nihon Kokenkogyo K.K.), 5 parts of a xylene solution (solid content, 50%) of an acrylic/polyol resin, 5 parts of xylene, and 1 part of an isocyanate hardener was applied by spraying to the reversibly thermochromic layer of the reversibly thermochromic light-transmitting laminate member produced in Example 2. Thus, a metallic glossy layer was formed. Thereafter, a topcoat ink obtained by evenly mixing 5 parts of a xylene solution (solid content, 50%) of an acrylic/polyol resin, 5 parts of xylene, 1 part of an isocyanate hardener, and 0.1 part of an ultraviolet absorber was applied to the metallic glossy layer by spraying. The resultant coating was dried at 70°C for 30 minutes to form a transparent topcoat layer.

[0074] When the laminate member thus obtained was cooled to 25°C or lower, the reversibly thermochromic composition assumed a blue color and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the interior of the minicar could be perceived through the metallic-blue body.

[0075] When the laminate member was heated to 35°C or higher, the reversibly thermochromic composition was decolored and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the interior of the minicar could be perceived through the transparent body.

[0076] The light transmittance of the metallic glossy layer itself was measured with a spectrophotometer (U-3210, manufactured by Hitachi Ltd.) and was found to be 40%. The light transmittances of the laminate member were 20% and 30% when the reversibly thermochromic composition was in a colored state and in a decolored state, respectively.

**EXAMPLE 5**

Production of Reversibly Thermochromic Light-Transmitting Laminate member with Metallic Luster:

[0077] A metallic luster ink obtained by evenly mixing 1 part of a metallic luster pigment (trade name, Iridion 225; manufactured by Merck & Co., Inc.), 5 parts of a xylene solution (solid content, 50%) of an acrylic/polyol resin, 5 parts of xylene, and 1 part of an isocyanate hardener was applied by spraying to the reversibly thermochromic layer of the reversibly thermochromic light-transmitting laminate member produced in Example 2. Thus, a metallic glossy layer was formed. Thereafter, a

topcoat ink obtained by evenly mixing 5 parts of a xylene solution (solid content, 50%) of an acrylic/polyol resin, 5 parts of xylene, 1 part of an isocyanate hardener, and 0.1 part of an ultraviolet absorber was applied to the metallic glossy layer by spraying. The resultant coating was dried at 70°C for 30 minutes to form a transparent topcoat layer.

[0078] When the laminate member thus obtained was cooled to 25°C or lower, the reversibly thermochromic composition assumed a blue color and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the interior of the minicar could be perceived through the metallic-blue body. When the laminate member was heated to 35°C or higher, the reversibly thermochromic composition was decolored and the reversibly thermochromic layer retained excellent transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the interior of the minicar could be perceived through the transparent body.

[0079] The light transmittance of the metallic glossy layer itself was measured with a spectrophotometer (U-3210, manufactured by Hitachi Ltd.) and was found to be 25%. The light transmittances of the laminate member were 10% and 20% when the reversibly thermochromic composition was in a colored state and in a decolored state, respectively.

**COMPARATIVE EXAMPLE 1**

[0080] A microencapsulated pigment was produced in the same manner as in Example 1, except that the particle diameter of the microencapsulated pigment in Example 1 was changed to 6.0 μm. Thereafter, a reversibly thermochromic material was prepared and a reversibly thermochromic laminate member was obtained, in the same manner as in Example 1.

[0081] When the laminate member thus obtained was heated to 36°C or higher, the reversibly thermochromic composition was decolored but the reversibly thermochromic layer had poor transparency. As a result, not only the feeling of transparency given by the substrate was impaired but also the silver color attributable to the paper coated with vapor-deposited aluminum was difficult to perceive. Namely, the laminate member was perceived as a poorly lustrous toy jewel giving a feeling of insufficient transparency.

**COMPARATIVE EXAMPLE 2**

[0082] A microencapsulated pigment was produced in the same manner as in Example 1, except that the particle diameter of the microencapsulated pigment in Example 1 was changed to 0.65 μm. Thereafter, a reversibly thermochromic material was prepared and a reversibly thermochromic laminate member was obtained, in the same manner as in Example 1.

**[0083]** The laminate member thus obtained had a low color density at the time when the reversibly thermochromic composition was in a colored state. The laminate member was hence further coated with the reversibly thermochromic material. As a result, the amount of the material necessary for obtaining the same color density as in Example 1 was about two times the amount used in Example 1. Furthermore, when the reversibly thermochromic material was allowed to stand for 3 hours and then applied to the same substrate, the resultant laminate member in a colored state had a reduced color density. It is thought that the microencapsulated pigment had deteriorated in the ink.

#### COMPARATIVE EXAMPLE 3

**[0084]** A reversibly thermochromic laminate member was obtained in the same manner as in Example 2, except that the polyurethane resin having an elongation at break of 800% used in Example 2 was replaced with a polyurethane resin having an elongation at break of 200%.

**[0085]** When the reversibly thermochromic laminate member thus obtained was cooled to 25°C or lower, the reversibly thermochromic composition assumed a blue color but the reversibly thermochromic layer had poor transparency and, hence, impaired the feeling of transparency given by the substrate. As a result, it was difficult to perceive the interior through the blue body.

#### COMPARATIVE EXAMPLE 4

**[0086]** A microencapsulated pigment was produced in the same manner as in Example 2, except that the particle diameter of the microencapsulated pigment in Example 2 was changed to 8.0 μm. Thereafter, a reversibly thermochromic material and a reversibly thermochromic laminate member were obtained in the same manner as in Example 2, except that the polyurethane resin having an elongation at break of 800% was replaced with a polyurethane resin having an elongation at break of 200%.

**[0087]** When the reversibly thermochromic laminate member thus obtained was cooled to 25°C or lower, the reversibly thermochromic composition assumed a blue color but the reversibly thermochromic layer had poor transparency and, hence, impaired the feeling of transparency given by the substrate. As a result, it was difficult to perceive the interior through the blue body.

**[0088]** Furthermore, when the laminate member was heated to 35°C or higher, the reversibly thermochromic composition was decolored but the reversibly thermochromic layer had poor transparency and, hence, impaired the feeling of transparency given by the substrate. As a result, the interior was difficult to perceive.

#### COMPARATIVE EXAMPLE 5

**[0089]** A reversibly thermochromic material was obtained by dissolving a reversibly thermochromic composition consisting of 1.5 parts of 1,2-benz-6-(N-ethyl-N-isobutylamino)fluoran as an electron-donating color-forming organic compound (a), 5 parts of 1,1-bis(4-hydroxyphenyl)-n-decane as an electron-accepting compound (b), and 15 parts of cetyl alcohol as a reaction medium (c) in 240 parts of an MIBK solution containing 30 parts of a vinyl chloride/vinyl acetate copolymer resin (composition: vinyl chloride/vinyl acetate = 86/14) and 30 parts of a vinyl chloride/vinyl acetate copolymer resin (composition: vinyl chloride/vinyl acetate/vinyl alcohol = 91/3/6).

**[0090]** A platy piece as a substrate was molded from an ABS resin. This substrate was coated with the reversibly thermochromic light-transmitting material by spraying, and the resultant coating was dried at 70°C for 60 minutes to form a reversibly thermochromic layer (thickness, 30 μm).

**[0091]** When the laminate member thus obtained was cooled to 20°C or lower, the reversibly thermochromic composition assumed a pink color and the reversibly thermochromic layer retained transparency and, hence, did not impair the feeling of transparency given by the substrate. As a result, the laminate member was pink and transparent. When this laminate member was heated to 43°C or higher, the reversibly thermochromic composition became colorless and the laminate member became transparent.

**[0092]** However, the reversibly thermochromic composition assumed almost no color when the laminate member was cooled after two-week standing. Namely, the function of changing in color had been impaired. It is presumed that the ABS resin serving as the substrate partly dissolved and exerted a chemical influence on the reversibly thermochromic composition to inhibit its color change function.

**[0093]** As described above, the invention can provide a reversibly thermochromic light-transmitting laminate member which has a satisfactory thermochromic function and undergoes alternate color changes between a colorless transparent state and a colored transparent state.

**[0094]** While the invention has been described in detail and with reference to specific embodiments thereof, it will be apparent to one skilled in the art that various changes and modifications can be made therein without departing from the spirit and scope thereof.

**[0095]** This application is based on Japanese patent applications No. 2000-368015 filed December 4, 2000 and No. 2001-316342 filed October 15, 2001, the entire contents thereof being hereby incorporated by reference.

**Claims**

1. A reversibly thermochromic light-transmitting laminate member which comprises a substrate and formed thereon a reversibly thermochromic layer comprising a transparent resin and fixedly dispersed therein a reversibly thermochromic microencapsulated pigment containing a reversibly thermochromic composition comprising (a) an electron-donating color-forming organic compound, (b) an electron-accepting compound, and (c) a reaction medium determining the temperatures at which the color reactions of component (a) with component (b) take place, the reversibly thermochromic microencapsulated pigment being a deformable pigment which has an average particle diameter of from 1.0 to 5.0  $\mu\text{m}$ , the transparent resin having a tensile elongation at break of 400% or higher.  
5  
10  
15
2. The reversibly thermochromic light-transmitting laminate member of claim 1, wherein the proportion of the reversibly thermochromic microencapsulated pigment to the transparent resin is from 1/0.5 to 1/20 in terms of weight ratio on a solid basis and the reversibly thermochromic layer has a thickness of from 1 to 100  $\mu\text{m}$ .  
20  
25
3. The reversibly thermochromic light-transmitting laminate member of claim 1, wherein the transparent resin is a resin selected from the group consisting of ethylene/vinyl acetate copolymer resins, polyurethane resins, and polyamide resins.  
30
4. The reversibly thermochromic light-transmitting laminate member of claim 1, which has, formed over the reversibly thermochromic layer, a metallic glossy layer containing a metallic luster pigment having transparency.  
35
5. The reversibly thermochromic light-transmitting laminate member of claim 4, wherein the metallic glossy layer has a light transmittance of 5% or higher.  
40

45

50

55

**THIS PAGE BLANK (USPTO)**

(19)



Europäisches Patentamt  
European Patent Office  
Office européen des brevets



(11)

**EP 1 211 300 A3**

(12)

**EUROPEAN PATENT APPLICATION**

(88) Date of publication A3:  
07.05.2003 Bulletin 2003/19

(51) Int Cl.7: **C09K 9/00, C09K 9/02**

(43) Date of publication A2:  
05.06.2002 Bulletin 2002/23

(21) Application number: 01128409.8

(22) Date of filing: 04.12.2001

(84) Designated Contracting States:  
**AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU  
MC NL PT SE TR**  
Designated Extension States:  
**AL LT LV MK RO SI**

(30) Priority: 04.12.2000 JP 2000368015  
15.10.2001 JP 2001316342

(71) Applicant: **THE PILOT INK CO., LTD.**  
Nagoya-shi, Aichi-ken (JP)

(72) Inventors:  
• Senga, Kuniyuki  
Showa-ku, Nagoya-shi, Aichi (JP)  
• Sugai, Jun  
Showa-ku, Nagoya-shi, Aichi (JP)  
• Koide, Shigehiro  
Showa-ku, Nagoya-shi, Aichi (JP)

(74) Representative: **HOFFMANN - EITLE**  
Patent- und Rechtsanwälte  
Arabellastrasse 4  
81925 München (DE)

(54) **Reversibly thermochromic light-transmitting laminate member**

(57) A reversibly thermochromic light-transmitting laminate member which has a satisfactory thermochromic function and undergoes alternate color changes between a colorless transparent state and a colored transparent state.

**EP 1 211 300 A3**



European Patent  
Office

## EUROPEAN SEARCH REPORT

Application Number  
EP 01 12 8409

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
P,X	EP 1 084 860 A (PILOT INK CO LTD) 21 March 2001 (2001-03-21) * page P.3, line 2-5 * * tables 1,2 * * examples 3,4,9 * * page 11, line 42-46 * * page 12, line 4-8 *	1-3	C09K9/00 C09K9/02
X	EP 0 789 066 A (PILOT INK CO LTD) 13 August 1997 (1997-08-13) * page 3, line 1-7 * * examples * * claims *	1-5	
X	US 4 681 791 A (SHIBAHASHI YUTAKA ET AL) 21 July 1987 (1987-07-21) * column 3, line 11 - column 4, line 20 * * table 1 *	1-5	
A	EP 0 919 604 A (PILOT INK CO LTD) 2 June 1999 (1999-06-02) * the whole document *	1-5	TECHNICAL FIELDS SEARCHED (Int.Cl.7) C09K
The present search report has been drawn up for all claims			
Place of search <b>MUNICH</b>		Date of completion of the search <b>14 March 2003</b>	Examiner <b>Doslik, N</b>
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			

EPO FORM 1503 03 82 (P04C01)

# ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 01 12 8409

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

14-03-2003

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 1084860	A	21-03-2001	CA 2318700 A1	17-03-2001
			EP 1084860 A2	21-03-2001
			JP 2001152041 A	05-06-2001
			US 6494950 B1	17-12-2002
			JP 2002012787 A	15-01-2002
-----				
EP 0789066	A	13-08-1997	JP 9208850 A	12-08-1997
			CA 2196513 A1	07-08-1997
			EP 0789066 A2	13-08-1997
			US 5879438 A	09-03-1999
-----				
US 4681791	A	21-07-1987	JP 1694611 C	17-09-1992
			JP 3057993 B	04-09-1991
			JP 61179389 A	12-08-1986
			JP 1914086 C	23-03-1995
			JP 6037746 B	18-05-1994
			JP 61179371 A	12-08-1986
			JP 1852269 C	21-06-1994
			JP 5063197 B	10-09-1993
			JP 61181481 A	14-08-1986
			JP 1852273 C	21-06-1994
			JP 5063198 B	10-09-1993
			JP 61196988 A	01-09-1986
			JP 61225369 A	07-10-1986
			AU 584163 B2	18-05-1989
			AU 5284586 A	07-08-1986
			CA 1240883 A1	23-08-1988
			DE 3602805 A1	31-07-1986
			FR 2576616 A1	01-08-1986
			GB 2170228 A ,B	30-07-1986
			KR 9209264 B1	15-10-1992
-----				
EP 0919604	A	02-06-1999	JP 2000072977 A	07-03-2000
			EP 0919604 A2	02-06-1999
			JP 11192800 A	21-07-1999
			US 6228804 B1	08-05-2001
-----				

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

**THIS PAGE BLANK (USPTO)**